167. Acetylene Reactions. Part II. Oxidation,* Hydration, and Mannich Reactions with 3-Aminobut-1-ynes.

By J. D. Rose and B. C. L. WEEDON.

Air oxidation of 3-alkyl- and -dialkyl-aminobut-1-ynes in presence of a copper catalyst gives high yields of 2:7-alkyl- and -dialkyl-amino-octa-3:5-diynes, which are catalytically hydrogenated to alkylated 2:7-diamino-octanes. Hydration of 3-diethylaminobut-1-yne with mercury salts gives 3-diethylaminobutan-2-one, and interaction of 3-diethylamino- and 3-morpholino-but-1-yne with formaldehyde and secondary amines affords acetylenic diamines.

RECENTLY published information from German sources on the reaction products of amines and acetylene (Rose, "Products formed by Interaction of Acetylene and Amines"; B.I.O.S. Final Report No. 359, Item No. 22, H.M.S.O. 1946; cf. Part I, preceding paper) have made the 3-mono- and -di-alkylaminobut-1-ynes readily available substances offering a fruitful field for research. This paper records some reactions which have been effected with this class of compound.

The hydration of 3-dialkylaminobut-1-ynes with mineral acids in the presence of mercury salts has been reported briefly as giving the corresponding amino-ketones (II) (I.G., B.P. 510,876), and this has been confirmed in the case of 3-diethylaminobut-1-yne (I; R = R' = Et), 3-diethylaminobutan-2-one (II; R = R' = Et) being obtained

Mannich-type reactions have been described with some monosubstituted acetylenes

* Patent application pending.

(Mannich and Chang, Ber., 1933, 66, 418; Jones, Marszak, and Bader, J., 1947, 1578), and, in the presence of a copper catalyst with acetylene (I.G., B.P. 510,904), vinylacetylene (Coffmann, J. Amer. Chem. Soc., 1935, 57, 1978; U.S.P. 2,136,177), but-3-yn-2-ol (Rose, "Preparation of 5-Diethylaminopentan-2-ol by the Reppe Process," B.I.O.S. Final Report No. 353, Item No. 22, H.M.S.O.) and diacetylene (Microfilm of F.D 3781/45, p. 000686).

3-Diethylaminobut-1-yne (I; R = R' = Et) reacted readily with paraformaldehyde and diethylamine or piperidine giving the acetylenic *diamines* (III; R = R' = R'' = R''' = Et, and R = R' = Et, $R''R''' = C_5H_{10}$) in 70 and 55% yields respectively; catalytic hydrogenation of these afforded the corresponding saturated 1:4-*diaminopentanes*.

The only product which could be isolated from diethylaminobutyne (I; R = R' = Et), paraformaldehyde, and morpholine was dimorpholinomethane, but 3-morpholinobut-1-yne, with paraformaldehyde and diethylamine gave the required 4-morpholino-1-diethylaminopent-2-yne (III; $RR' = C_4H_8O$, R'' = R''' = Et).

RR'N•CHMe•C : CH	RR'N•CHMe•COMe	RR'N·CHMe•C : C·CH ₂ ·NR''R'''
(I.)	(II.)	(III.)

The aerial oxidation in the presence of cuprous salts of compounds containing the ethynyl group, -C:CH, to give diacetylenes, -C:CC:C-, has previously been studied with the acetylenic hydrocarbons (Zal'kind and Fundyler, Ber., 1936, **69**, 128; J. Gen. Chem. Russia, 1939, **9**, 1725) and acetylenic alcohols (Zal'kind et al., ibid., 1937, **31**, 4283; 1939, **9**, 971, 1725; Bowden, Heilbron, Jones, and Sargent, J., 1947, 1579), and that this method can be used as a highly efficient industrial process has been demonstrated by the I.G. in the oxidation of propargyl alcohol to hexa-2: 4-diyne-1: 6-diol (Rose, "Propargyl Alcohol; Dehydration and Oxidation to Hexadiynediol," B.I.O.S. Final Report No. **357**, Item No. **22**, H.M.S.O.). This reaction has now been applied to the aminobutynes (I), and it was found that oxidation was most conveniently effected by using an aqueous solution of the amine hydrochloride in the presence of only a catalytic amount of cuprous chloride. Under these conditions, a homogeneous solution was obtained, and the deposition of cuprous oxide with loss of catalyst, observed when the free

(IV.) R'RN•CHMe•C**:**C•C**:**C•CHMe•NRR' Et₂N•CH₂•CH:CH•C**:**C•C**:**C•CH:CH•CH₂•NEt₂ (V.)

base is employed, was avoided. In this manner the diacetylenic diamines (IV; R = R' = Et; $RR' = C_4H_8O$; $R = Pr^i$, R' = H; R = Bu, R' = H) were prepared from the corresponding aminobutynes (I) in yields of 66—95%. A similar oxidation of 1-diethylaminopent-2-en-4-yne gave 1:10-bisdiethylaminodeca-2:8-diene-4:6-diyne (V) in 65% yield.

The light-absorption properties of the diacetylenic diamines (see table) are in good agreement with those of the corresponding diacetylenic glycols.

$\begin{array}{c} Compound.\\ (Et_2N \cdot CHMe \cdot C \overset{\circ}{\cdot} C)_2 \end{array} .$	$\lambda_{\max. A.} \ {}^{A.2420} \ {}^{2550} \$	ε _{max.} . 1,200 720	Compound. (CH ₃ ·CH[OH]·C:C) ₂ (1)	$\lambda_{\max. A.} = 2300 \\ 2370$	ε _{max.} . 1,930 1,860
$(O < [CH_2 \cdot CH_2]_2 > N \cdot CHMe \cdot C \vdots C)_2 \dagger$	2420 * 2505 * 2570 *	$875 \\ 450 \\ 365$	$(Me_2C \cdot [OH] \cdot C:C)_2 (1) \dots$	$2290 \\ 2410 \\ 2560$	31 0 3 00 1 80
(Pr ⁱ NH·CHMe·C : C) ₂ ,H ₂ O	$\begin{array}{c} 2420 \\ 2570 \end{array}$	630 330	(HO·CH ₂ ·CH:CH·C:C) ₂ (2)	$2300 \\ 2370 \\ 2470$	38 ,000 3 9,000 3 0,000
$(\mathrm{Bu^tNH}\text{\cdot}\mathrm{CHMe}\text{\cdot}\mathrm{C}\text{\cdot}\mathrm{C})_{2} \dots \dots \dots$	2470 * 2560 *	1,550 1,000		2620 2765 2930	12,000 20,000 30,000
(Et ₂ N·CH ₂ ·CH:CH·C:C) ₂	2290 2390 2490 2640 2790 2960 3150	26,000 24,500 19,000 9,500 15,000 24,500 22,000		3120	23,000
* Indicator on	inflorion		+ In mothenel		

Indicates an inflexion

† In methanol.

Bowden, Heilbron, Jones, and Sargent (*loc. cit.*).
 Heilbron, Jones, and Sondheimer (*J.*, 1947, 1586).

It has been shown (Loritsch and Vogt, J. Amer. Chem. Soc., 1939, 61, 1462) that aniline reacts with monoalkylacetylenes, with a boron trifluoride-mercuric oxide catalyst, to give the

anil of the corresponding methyl alkyl ketone. Treatment of 3-diethylaminobut-1-yne (I; R = R' = Et) with aniline under similar conditions gave diacetyl dianil (VIII).

$$\begin{array}{cccc} \text{Et}_{2}\text{N}\text{\cdot}\text{CHMe}\text{\cdot}\text{C}(\text{:NPh})\text{Me} & \longrightarrow & \text{Ph}\text{\cdot}\text{NH}\text{\cdot}\text{CHMe}\text{\cdot}\text{C}(\text{:NPh})\text{Me} & \xrightarrow{\text{Hs}O} & \text{Ph}\text{\cdot}\text{N}\text{\cdot}\text{CMe}\text{\cdot}\text{C}(\text{:NPh})\text{Me} \\ & (\text{VI.}) & (\text{VIII.}) & (\text{VIII.}) \end{array}$$

IJ~O

The reaction probably depended on the intermediate formation of (VI) and (VII), the latter then being oxidised by mercuric oxide (the formation of mercury during the reaction was observed).

EXPERIMENTAL.

Analyses are by Mr. E. S. Morton. All m. p.s are uncorrected.

Light-absorption measurements were carried out in ethanol except where stated otherwise.

Light-absorption measurements were carried out in ethanol except where stated otherwise. 3-Diethylaminobutan-2-one.—Mercuric sulphate (0·2 g.) was added to a solution of 3-diethylamino-but-1-yne (5 g.) in sulphuric acid (d 1·84; 12 c.c.) and water (3 c.c.). The mixture was heated on the steam-bath for 1 hour, cooled, and poured on ice. Excess of sodium hydroxide solution (40% w/v) was added to the solution, and the precipitated oil extracted with ether. Evaporation of the ethereal solution and distillation of the residue gave 3-diethylaminobutan-2-one (4·5 g.), b. p. 56—58°/10 mm., n_D^{17} 1·4303 (Found: C, 67·0; H, 11·65; N, 9·7. Calc. for C₈H₁₇ON: C, 67·1; H, 11·8; N, 9·8%)) (I.G., B.P. 510,876, give b. p. 168°). An aqueous alcoholic solution of the amino-ketone and semicarbazide acetate was left overnight. The semicarbazone was precipitated on addition of excess of alkali; it crystallised from alcohol in needles, m. p. 189·5—190·5° (Found: C, 54·25; H, 9·8; N, 27·8. C₉H₂₀ON₄ requires C, 54·0; H, 10·05; N, 28·0%). 1 : 4-Bisdiethylaminopent-2-yne (11; R = R' = R'' = R''' = Et).—A mixture of 3-diethylamino-but-1-yne (12·5 g.), diethylamine (8 g.), paraformaldehyde (4 g.), and dioxan (20 c.c.) was heated on the steam-bath for 13 hours, cooled, and diluted with water (400 c.c.). The product was extracted with ether, the ethereal solution evaporated, and the residue distilled, giving 1 : 4-bisdiethylaminopent-2-yne

steam-bath for 13 hours, cooled, and diluted with water (400 c.c.). The product was extracted with ether, the ethereal solution evaporated, and the residue distilled, giving 1: 4-bisdiethylaminopent-2-yne (15·3 g.), b. p. 104—106°/8 mm., n_{15}^{10} ; 1:4598 (Found : C, 74·4; H, 12·15. C₁₃H₂₆N₂ requires C, 74·2; H, 12·45%). The dihydrochloride, m. p. 205—206°, crystallised from alcohol-ether (Found : C, 55·25; H, 9·75; N, 9·6. C₁₃H₂₆N₂,2HCl requires C, 55·1; H, 9·95; N, 9·9%). The dipicrate, m. p. 167—168·5°, crystallised in prisms from 2-ethoxyethyl alcohol (Found : C, 44·95; H, 4·75; N, 16·35. C₁₃H₂₆N₂,2C₆H₃O₂N₃ requires C, 44·9; H, 4·8; N, 16·75%). The dimethiodide, m. p. 168°, was prepared in acetone solution and purified from alcohol-ether (Found : C, 36·45; H, 6·65; N, 5·55%). 1: 4-Bisdiethylaminopentae.—A solution of 1: 4-bisdiethylaminopent-2-yne (4 g.) in methanol (50 c.c.) was shaken with hydrogen and Raney nickel until absorption was complete. After filtration,

(50 c.c.) was shaken with hydrogen and Raney nickel until absorption was complete. After filtration,

(50 c.c.) was shaken with hydrogen and Raney nickel until absorption was complete. After filtration, acidification, and evaporation of the solution, excess of sodium hydroxide (20%) was added to the residue, and the product isolated with ether. Distillation gave 1: 4-bisdiethylaminopentane (2 g.), b. p. 105°/10 mm., n_1^{15} 1.4446 (Found : C, 73·2; H, 13·95; N, 13·2. $C_{13}H_{30}N_2$ requires C, 72·85; H, 14·1; N, 13·05%). The dipicrate, m. p. 192—194°, crystallised in needles from 2-ethoxyethyl alcohol (Found : C, 44·65; H, 54; N, 16·65. $C_{13}H_{20}N_2.2C_6H_3O_7N_3$ requires C, 44·6; H, 5·05; N, 16·9%). 1-Piperidino-4-diethylaminopent-2-yne (III; R = R' = Et, R''R''' = C_5H_{10}).—3-Diethylaminobut-1-yne (12·5 g.), piperidine (8·5 g.), paraformaldehyde (4 g.), and dioxan (20 c.c.) were heated on the steam-bath for 13 hours, and the product isolated as usual. Distillation gave 1-piperidino-4-diethyl-aminopent-2-yne (11·5 g.), b. p. 132—136°/9 mm., n_1^{16} 1.4810 (Found : C, 75·75; H, 11·8.%). The dipicrate, crystallised in needles from alcohol-ether, had m. p. 200—202° (decomp.) (Found : C, 56·55; N, 9·35; N, 9·35, C.14H_26N_2.2HCl requires C, 56·95; H, 9·55; N, 9·5%). The dipicrate, m. p. 190—191° (decomp.), crystallised in needles from acetone (Found : C, 46·1; H, 4·6. $C_{14}H_{26}N_2.2C_6H_3O_7N_3$ requires C, 45·85; H, 4·75%). The dimethiodide, m. p. 198—201° (decomp.), was crystallised from alcohol-ether (Found : C, 37·6; H, 6·0; N, 5·5. $C_{16}H_{32}N_2I_2$ requires C, 37·95; H, 6·35, N, 9·55%). 6·35, N, 5·55%).

1-Piperidino-4-diethylaminopentane.—A solution of 1-piperidino-4-diethylaminopent-2-yne (2.25 g.) in methanol was shaken with hydrogen and Raney nickel until absorption was complete. Isolation of The product gave 1-*piperidino*-4-*diethylaminopentane* (1.3 g.), b. p. 130–134°/12 mm., n_D^{17} , 1-4644 (Found : N, 12·1. $C_{14}H_{30}N_2$ requires N, 12·35%). The *dipicrate*, m. p. 177–178°, crystallised as needles from aqueous acetone (Found : C, 45·6; H, 5·25; N, 16·5. $C_{14}H_{36}N_2, 2C_6H_3O_7N_3$ requires C, 45·35; H, 5·25; N, 16·3%).

4-Morpholino-1-diethylaminopent-2-yne (III; $RR' = C_4H_8O$, R'' = R''' = Et).—A mixture of 4-Morpholino-1-diethylaminopent-2-yne (III; $RR' = C_4H_8O$, R'' = R'' = Et).—A mixture of 3-morpholinobut-1-yne (13.9 g.), diethylamine (8 g.), paraformaldehyde (4 g.), and dioxan (20 c.c.) was heated on the steam-bath for 15 hours. Isolation of the product gave 4-morpholino-1-diethylaminopent-2-yne (18.5 g.), b. p. 86—90°/0·1 mm., n_{22}^{22} 1·4800 (Found : N, 12·45. $C_{13}H_{24}ON_2$ requires N, 12·5%). The dipicrate, m. p. 204—206°, formed needles from glycol monomethyl ether (Found : C, 44·3; H, 4·45; N, 16·35. $C_{13}H_{24}ON_2, 2C_8H_3O_{7N_3}$ requires C, 44·0; H, 4·45; N, 16·45%). 2 : 7-Bisdiethylamino-octa-3 : 5-diyne (IV; R = R' = Et).—A solution of 3-diethylaminobut-1-yne (75 g.) in dilute hydrochloric acid (2N; 310 c.c.) was added to a mixture of cuprous chloride (15 g.), ammonium chloride (45 g.), and water (53 c.c.). The resulting solution was warmed to 50—55° and stirred vigorously, and air was bubbled through it until green salts began to be deposited (9 hours).

The mixture was cooled and filtered, and excess of concentrated ammonia solution added to the filtrate. The precipitated oil was extracted with ether, the ethereal solution evaporated, and the residue distilled, giving 2:7-bisdiethylamino-octa-3:5-diyne (68 g.), b. p. $130-134^{\circ}/0.2$ mm., $n_{\rm D}^{\rm B^*}$ 1.4958 (Found : C, 77.35; H, 11.25; N, 11.25. $C_{16}H_{28}N_2$ requires C, 77.3; H, 11.35; N, 11.3%). The dipicrate, m. p. 230° (decomp.) (bath preheated to 190°), crystallised as leaflets from 2-ethoxyethyl alcohol (Found : C, 47.8; H, 4.85; N, 16.3. C₁₆H₂₈N₂,2C₆H₃O₇N₃ requires C, 47.6; H, 4.85; N, 15.9%).
2: 7-Bisdiethylamino-octane.-2: 7-Bisdiethylamino-octa-5: 7-diyne (4.5 g.) in methanol (50 c.c.)

2:7-Bisdiethylamino-octane. 2:7-Bisdiethylamino-octa-5:7-diyne (4.5 g.) in methanol (50 c.c.) was hydrogenated (Raney nickel), giving 2:7-bisdiethylamino-octane (2 g.), b. p. 83–87°/0·1 mm., n_D^{20} 1·4612 (Found : N, 11·15; equiv., 128·2. $C_{16}H_{36}N_2$ requires N, 10·9%; equiv., 128·2). The dipicrate, m. p. 186–190° (decomp.), crystallised from 2-ethoxyethyl alcohol as leaflets (Found : C, 47·2; H, 5·6. $C_{16}H_{36}N_2, 2C_{6}H_3O_7N_3$ requires C, 47·05; H, 5·9%).

III. p. 180–180 (decomp.), crystallised noise to be considered in the construction of the construction of

The *uthyurochoride*, featers from aqueous accorde, fait in. p. 242 (decomp.) (Found 1. 1., 5.1.) $C_{16}H_{24}O_{2}N_{2.}2$ HCl requires N, 8.0%). 2:7-Diisopropylamino-octa-3:5-diyne (IV; R = Pr¹; R' = H).—A solution of 3-isopropylaminobut-1-yne (100 g.) in dilute hydrochloric acid (2N; 440 c.c.) was added to a mixture of cuprous chloride (20 g.), ammonium chloride (60 g.), and water (70 c.c.), and the oxidation carried out as usual. Isolation gave 2: 7-diisopropylamino-octa-3: 5-diyne monohydrate (105 g.), m. p. 61°, which crystallised as needles from aqueous methanol (Found : C, 70·7; H, 10·95; N, 11·95. C₁₄H₂₄N₂, H₂O requires C, 70·5; H, 11·0; N, 11·75%). On standing in a desiccator this solid gives an oil, presumably the anhydrous diamine, which on exposure to the atmosphere is rapidly reconverted into the original hydrate. The dipicrate, m. p. 198—200° (decomp.), crystallised in prismatic needles from aqueous methanol (Found : C, 45·85; H, 4·4; N, 16·7. C₁₄H₂₄N₂, 2C₆H₃O₇N₃ requires C, 46·05; H, 4·45; N, 16·5%). The dihydrochloride, purified from alcohol-ether, had m. p. 242·5—243·5° (Found : C, 56·95; H, 8·7; N, 9·25. C₁₄H₂₄N₂, 2HCl requires C, 57·35; H, 8·95; N, 9·55%). 2:7-Diisopropylamino-octane.—A solution of 2:7-diisopropylamino-octa-3:5-diyne monohydrate (92 g.) in methanol (800 c.c.) was hydrogenated over Raney nickel until absorption was complete

2:7-Diisopropylamino-octane.—A solution of 2:7-diisopropylamino-octa-3:5-diyne monohydrate (92 g.) in methanol (800 c.c.) was hydrogenated over Raney nickel until absorption was complete. Isolation of the product gave 2:7-diisopropylamino-octane (71 g.), b. p. 92–95°/05 mm., n_{20}^{20} 1:435 (Found: N, 12·2. $C_{14}H_{32}N_2$ requires N, 12·25%). The dihydrochloride, crystallised from alcohol-ether, had m. p. 270° (decomp.) (Found: C, 55.65; H, 11·25. $C_{14}H_{32}N_2$,2HCl requires C, 55.8; H, 11·35%). 2:7-Di-n-butylamino-octa-3:5-diyne (IV; R = Bu; R' = H).—A solution of 3-n-butylaminobutlawne (II g.) in dilute budrochloric acid (2N: 45 c.) was added to a mixture of currous chloride (2 g.)

2 : 7-Di-n-butylamino-octa-3 : 5-diyne (IV; $\mathbf{R} = \mathbf{Bu}$; $\mathbf{R}' = \mathbf{H})$.—A solution of 3-n-butylaminobut-1-yne (11 g.) in dilute hydrochloric acid (2x; 45 c.c.) was added to a mixture of cuprous chloride (2 g.), ammonium chloride (6 g.), and water (7 c.c.). Aerial oxidation (14 hours) at 50° and isolation of the product gave 2 : 7-di-n-butylamino-octa-3 : 5-diyne (7-2 g.), b. p. 128—132°/0·06 mm., n_{23}^{24} 1-4952 (Found : N, 11-4. C₁₆H₂₈N₂ requires N, 11·3%). The di-a-naphthylurea, m. p. 163—164°, crystallised in micro-needles from aqueous alcohol (Found : C, 77·7; H, 7·1; N, 9·25. C₃₈H₄₂O₂N₄ requires C, 77·75; H, 7·2; N, 9·55%).

1: 10-Bisdiethylaminodeca-2: 8-diene-4: 6-diyne (V).—A solution of 1-diethylaminopent-2-en-4-yne (4 g.; prepared from pent-2-en-4-yn-1-ol by chlorination with thionyl chloride and then treatment of the chloride with diethylamine, cf. Jones, Lacey, and Smith, J., 1946, 940) in dilute hydrochloric acid (2N, 14 c.c.) was added to a mixture of cuprous chloride (0.65 g.), ammonium chloride (1.9 g.), and water (2.25 c.c.). Aerial oxidation (5.5 hours) and isolation of the product gave bisdiethylaminodeca-2: 8-diene-4: 6-diyne (2.6 g.), b. p. 155—160°/0·1 mm., n_{20}^{20} 1-5630 (Found: C, 78-6; H, 9.8. $C_{18}H_{28}N_2$ requires C, 79-3; H, 10·3%). The dipicrate, m. p. 146—148°, crystallised in prisms from aqueous acetone (Found: C, 49-0; H, 4.85; N, 15·3. $C_{18}H_{28}N_2, C_{26}H_{30}C_{3}$ requires C, 49-3; H, 4.7; N, 15·35%). The dimethiodide, needles from alcohol, had m. p. 211° (decomp.) (Found: C, 43·2; H, 6·1; N, 4.85. $C_{20}H_{34}N_2I_2$ requires C, 43·15; H, 6·15; N, 5·05%). The dihydrochloride, m. p. 215° (decomp.), crystallised as needles from alcohol (Found: N, 8·25. $C_{18}H_{28}N_2,2HCl$ requires N, 8·1%). Light absorption in methanol: maxima 2380, 2490, 2775, 2970, 3160; ε_{max} , 28,500, 26,000, 9,500, 16,500, 27,000 and 23,000 respectively.

Diacetyl Dianil (VIII).—A mixture of aniline (45 g.), mercuric oxide (15 g.), and boron trifluorideether complex (6 g.) was warmed to 50°, and 3-diethylaminobut-1-yne (60 g.) added dropwise (0.5 hour) with rapid stirring. The mixture was stirred at 50° for a further 4 hours, cooled, and filtered. Distillation of the filtrate gave 3-diethylaminobut-1-yne (30 g.), aniline (15 g.), and a residue which on trituration with methanol solidified. Recrystallisation from the same solvent gave diacetyl dianil (17 g.) as yellow plates, m. p. 138° undepressed on admixture with an authentic specimen (von Pechmann, Ber., 1888, **21**, 1415, gives m. p. 139°) (Found : N, 12·0. Calc. for $C_{16}H_{16}N_2$: N, 11·85%).

The authors thank Professor E. R. H. Jones for a sample of pent-2-en-4-yn-1-ol, and Dr. E. A. Braude for determining some of the absorption spectra.

RESEARCH LABORATORIES,

IMPERIAL CHEMICAL INDUSTRIES LTD. (DYESTUFFS DIVISION), BLACKLEY, MANCHESTER, 9.

[Received, September 3rd, 1948.]