# 167. Acetylene Reactions. Part II. Oxidation,* Hydration, and Mannich Reactions with 3-Aminobut-1-ynes. 


#### Abstract

By J. D. Rose and B. C. L. Weedon. Air oxidation of 3 -alkyl- and -dialkyl-aminobut-1-ynes in presence of a copper catalyst gives high yields of 2:7-alkyl- and -dialkyl-amino-octa-3:5-diynes, which are catalytically hydrogenated to alkylated 2:7-diamino-octanes. Hydration of 3-diethylaminobut-1-yne with mercury salts gives 3 -diethylaminobutan-2-one, and interaction of 3 -diethylamino- and 3 -morpholino-but-1-yne with formaldehyde and secondary amines affords acetylenic diamines.


Recently published information from German sources on the reaction products of amines and acetylene (Rose, "Products formed by Interaction of Acetylene and Amines"; B.I.O.S. Final Report No. 359, Item No. 22, H.M.S.O. 1946; cf. Part I, preceding paper) have made the 3 -mono- and -di-alkylaminobut-1-ynes readily available substances offering a fruitful field for research. This paper records some reactions which have been effected with this class of compound.

The hydration of 3-dialkylaminobut-1-ynes with mineral acids in the presence of mercury salts has been reported briefly as giving the corresponding amino-ketones (II) (I.G., B.P. 510,876), and this has been confirmed in the case of 3-diethylaminobut-1-yne ( $\mathrm{I} ; \mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}$ ), 3-diethylaminobutan-2-one (II; $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}$ ) being obtained

Mannich-type reactions have been described with some monosubstituted acetylenes

* Patent application pending.
(Mannich and Chang, Bev., 1933, 66, 418; Jones, Marszak, and Bader, J., 1947, 1578), and, in the presence of a copper catalyst with acetylene (I.G., B.P. 510,904), vinylacetylene (Coffmann, J. Amer. Chem. Soc., 1935, 57, 1978 ; U.S.P. 2,136,177), but-3-yn-2-ol (Rose, " Preparation of 5-Diethylaminopentan-2-ol by the Reppe Process," B.I.O.S. Final Report No. 353, Item No. 22, H.M.S.O.) and diacetylene (Microfilm of F.D 3781/45, p. 000686).

3-Diethylaminobut-1-yne ( $\mathrm{I} ; \mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}$ ) reacted readily with paraformaldehyde and diethylamine or piperidine giving the acetylenic diamines (III; $R=R^{\prime}=R^{\prime \prime}=R^{\prime \prime \prime}=E t$, and $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}, \mathrm{R}^{\prime \prime} \mathrm{R}^{\prime \prime \prime}=\mathrm{C}_{5} \mathrm{H}_{10}$ ) in 70 and $55 \%$ yields respectively; catalytic hydrogenation of these afforded the corresponding saturated 1:4-diaminopentanes.

The only product which could be isolated from diethylaminobutyne ( $I$; $R=R^{\prime}=E t$ ), paraformaldehyde, and morpholine was dimorpholinomethane, but 3 -morpholinobut-1-yne, with paraformaldehyde and diethylamine gave the required 4-morpholino-1-diethylaminopent-2-yne (III; $\mathrm{RR}^{\prime}=\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}, \mathrm{R}^{\prime \prime}=\mathrm{R}^{\prime \prime \prime}=\mathrm{Et}$ ).

## $\mathrm{RR}^{\prime} \mathrm{N} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \mathrm{CH}$ <br> (I.) <br> $\mathrm{RR}^{\prime} \mathrm{N} \cdot \mathrm{CHMe} \cdot \mathrm{COMe}$ <br> (II.) <br> $\mathrm{RR}^{\prime} \mathrm{N} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \mathrm{C} \cdot \mathrm{CH}_{2} \cdot \mathrm{NR}^{\prime \prime} \mathrm{R}^{\prime \prime \prime}$ <br> (III.)

The aerial oxidation in the presence of cuprous salts of compounds containing the ethynyl group, $-\mathrm{C}: \mathrm{CH}$, to give diacetylenes, $-\mathrm{C}: \mathrm{C} \cdot \mathrm{C}: \mathrm{C}-$, has previously been studied with the acetylenic hydrocarbons (Zal'kind and Fundyler, Ber., 1936, 69, 128; J. Gen. Chem. Russia, 1939, 9, 1725 ) and acetylenic alcohols (Zal'kind et al., ibid., 1937, 31, 4283; 1939, 9, 971, 1725; Bowden, Heilbron, Jones, and Sargent, J., 1947, 1579), and that this method can be used as a highly efficient industrial process has been demonstrated by the I.G. in the oxidation of propargyl alcohol to hexa-2 : 4-diyne-1 : 6-diol (Rose, " Propargyl Alcohol; Dehydration and Oxidation to Hexadiynediol," B.I.O.S. Final Report No. 357, Item No. 22, H.M.S.O.). This reaction has now been applied to the aminobutynes (I), and it was found that oxidation was most conveniently effected by using an aqueous solution of the amine hydrochloride in the presence of only a catalytic amount of cuprous chloride. Under these conditions, a homogeneous solution was obtained, and the deposition of cuprous oxide with loss of catalyst, observed when the free

## (IV.) $\mathrm{R}^{\prime} \mathrm{RN} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \mathrm{C} \cdot \mathrm{C}: \mathrm{C} \cdot \mathrm{CHMe} \cdot \mathrm{NRR}^{\prime} \quad \mathrm{Et}_{2} \mathrm{~N} \cdot \mathrm{CH}_{2} \cdot \mathrm{CH}: \mathrm{CH} \cdot \mathrm{C}: \mathrm{C} \cdot \mathrm{C}: \mathrm{C} \cdot \mathrm{CH}: \mathrm{CH}^{\prime} \cdot \mathrm{CH}_{2} \cdot \mathrm{NEt}_{2} \quad$ (V.)

base is employed, was avoided. In this manner the diacetylenic diamines (IV; $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}$; $\mathrm{RR}^{\prime}=\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O} ; \mathrm{R}=\mathrm{Pr}^{\mathrm{i}}, \mathrm{R}^{\prime}=\mathrm{H} ; \mathrm{R}=\mathrm{Bu}, \mathrm{R}^{\prime}=\mathrm{H}$ ) were prepared from the corresponding aminobutynes ( I ) in yields of $66-95 \%$. A similar oxidation of 1-diethylaminopent-2-en-4-yne gave 1:10-bisdiethylaminodeca-2:8-diene-4:6-diyne (V) in $65 \%$ yield.

The light-absorption properties of the diacetylenic diamines (see table) are in good agreement with those of the corresponding diacetylenic glycols.

| Compound. $\left(\mathrm{Et}_{2} \mathrm{~N} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \stackrel{\bullet}{\mathrm{C}}\right)_{2} \quad \ldots \ldots \ldots \ldots \ldots .$ | $\begin{gathered} \lambda_{\max .} \mathrm{A} . \\ 2420 \\ 2550 \end{gathered}$ | $\begin{aligned} & \varepsilon_{\text {max. }} \\ & 1,200 \\ & 720 \end{aligned}$ | Compound. $\left(\mathrm{CH}_{3} \cdot \mathrm{CH}[\mathrm{OH}] \cdot \mathrm{C}: \mathrm{C}\right)_{2}(\mathrm{l})$ | $\lambda_{\text {max. }}$ A. 2300 2370 | $\begin{aligned} & \varepsilon_{\text {max. } .} \\ & 1,930 \\ & 1,860 \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\left(\mathrm{O}<\left[\mathrm{CH}_{2} \cdot \mathrm{CH}_{2}\right]_{2}>\mathrm{N} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \mathrm{C}\right)_{2} \dagger$ | 2420 * | 875 | $\left(\mathrm{Me} 2_{2} \mathrm{C} \cdot[\mathrm{OH}] \cdot \mathrm{C}: \mathrm{C}\right)_{2}(\mathrm{l})$ | 2290 | 310 |
|  | 2505 * | 450 |  | 2410 | 300 |
|  | 2570 * | 365 |  | 2560 | 180 |
| $\left(\mathrm{Pr}^{1} \mathrm{NH} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \mathrm{C}_{2}{ }_{2}, \mathrm{H}_{2} \mathrm{O}\right.$ | 2420 | 630 | $\left(\mathrm{HO} \cdot \mathrm{CH}_{2} \cdot \mathrm{CH}: \mathrm{CH} \cdot \mathrm{C}: \mathrm{C}\right)_{2}(2)$ | 2300 | 38,000 |
|  | 2570 | 330 |  | 2370 | 39,000 |
|  |  |  |  | 2470 | 30,000 |
| $\left(\mathrm{Bu}{ }^{\mathrm{N}} \mathrm{H} \cdot \mathrm{CHMe} \cdot \mathrm{C}: \mathrm{C}\right)_{2}$ | 2470 * | 1,550 |  | 2620 | 12,000 |
|  | 2560 * | 1,000 |  | 2765 | 20,000 |
|  |  |  |  | 2930 | 30,000 |
| $\left(\mathrm{Et}_{2} \mathrm{~N} \cdot \mathrm{CH}_{2} \cdot \mathrm{CH}: \mathrm{CH} \cdot \mathrm{C}: \mathrm{C}\right)_{2}$ | 2290 | 26,000 |  | 3120 | 23,000 |
|  | 2390 | 24,500 |  |  |  |
|  | 2490 | 19,000 |  |  |  |
|  | 2640 | 9,500 |  |  |  |
|  | 2790 | 15,000 |  |  |  |
|  | 2960 | 24,500 |  |  |  |
|  | 3150 | 22,000 |  |  |  |
| * Indicates an inflexion. $\dagger$ In methanol |  |  |  |  |  |
| (1) Bowd <br> (2) Heilbr | n, Heilb ron, Jone | on, Jon and S | nd Sargent (loc. cit.). eimer ( $J ., 1947,1586$ ). |  |  |

It has been shown (Loritsch and Vogt, J. Amer. Chem. Soc., 1939, 61, 1462) that aniline reacts with monoalkylacetylenes, with a boron trifluoride-mercuric oxide catalyst, to give the
anil of the corresponding methyl alkyl ketone. Treatment of 3-diethylaminobut-1-yne ( $I ; R=R^{\prime}=E t$ ) with aniline under similar conditions gave diacetyl dianil (VIII).


The reaction probably depended on the intermediate formation of (VI) and (VII), the latter then being oxidised by mercuric oxide (the formation of mercury during the reaction was observed).

## Experimental.

## Analyses are by Mr. E. S. Morton. All m. p.s are uncorrected.

Light-absorption measurements were carried out in ethanol except where stated otherwise.
3-Diethylaminobutan-2-one.-Mercuric sulphate ( $0 \cdot 2 \mathrm{~g}$.) was added to a solution of 3 -diethylamino-but-1-yne ( 5 g .) in sulphuric acid ( $1.84 ; 12 \mathrm{c.c}$.) and water ( $3 \mathrm{c} . \mathrm{c}$.). The mixture was heated on the steam-bath for 1 hour, cooled, and poured on ice. Excess of sodium hydroxide solution ( $40 \% \mathrm{w} / \mathrm{v}$ ) was added to the solution, and the precipitated oil extracted with ether. Evaporation of the ethereal solution and distillation of the residue gave 3 -diethylaminobutan- 2 -one ( 4.5 g.$)$, b. p. $56-58^{\circ} / 10 \mathrm{~mm}$., $n_{\mathrm{D}}^{17^{\circ}} 1.4303$ (Found: C, $67 \cdot 0$; $\mathrm{H}, 11 \cdot 65$; N, $9 \cdot 7$. Calc. for $\mathrm{C}_{8} \mathrm{H}_{17} \mathrm{ON}$ : C, $67 \cdot 1 ; \mathrm{H}, 11.8 ; \mathrm{N}, 9 \cdot 8 \%$ ) (I.G., B.P. 510,876 , give b. p. $168^{\circ}$ ). An aqueous alcoholic solution of the amino-ketone and semicarbazide acetate was left overnight. The semicarbazone was precipitated on addition of excess of alkali; it crystallised from alcohol in needles, m. p. $189 \cdot 5-190 \cdot 5^{\circ}$ (Found : C, $54 \cdot 25 ; \mathrm{H}, 9 \cdot 8$; N, 27.8. $\mathrm{C}_{9} \mathrm{H}_{20} \mathrm{ON}_{4}$ requires $\mathrm{C}, 54 \cdot 0 ; \mathrm{H}, 10.05 ; \mathrm{N}, 28.0 \%$ ).
$\mathrm{l}: 4$-Bisdiethylaminopent-2-yne (III; $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{R}^{\prime \prime}=\mathrm{R}^{\prime \prime \prime}=\mathrm{Et}$ ). -A mixture of 3 -diethylamino-but-1-yne ( 12.5 g .), diethylamine ( 8 g .), paraformaldehyde ( 4 g .), and dioxan ( 20 c.c.) was heated on the steam-bath for 13 hours, cooled, and diluted with water ( 400 c.c.). The product was extracted with ether, the ethereal solution evaporated, and the residue distilled, giving 1:4-bisdiethylaminopent-2-yne ( $15 \cdot 3 \mathrm{~g}$.), b. p. $104-106^{\circ} / 8 \mathrm{~mm}$., $n_{\mathrm{D}}^{18 \mathrm{~B}} 1.4598$ (Found: C, $74 \cdot 4 ; \mathrm{H}, 12 \cdot 15 . \mathrm{C}_{13} \mathrm{H}_{26} \mathrm{~N}_{2}$ requires $\mathrm{C}, 74 \cdot 2$; $\mathrm{H}, 12.45 \%$ ) . The dihydrochloride, m. p. 205-206 ${ }^{\circ}$, crystallised from alcohol-ether (Found : C, $55 \cdot 25$; $\mathrm{H}, 9.75 ; \mathrm{N}, 9 \cdot 6$. $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{~N}_{2}, 2 \mathrm{HCl}$ requires $\mathrm{C}, 55 \cdot 1 ; \mathrm{H}, 9.95 ; \mathrm{N}, 9.9 \%$ ). The dipicrate, m. p. $167-168.5^{\circ}$, crystallised in prisms from 2-ethoxyethyl alcohol (Found: C, $44.95 ; \mathrm{H}, 4.75$; $\mathrm{N}, 16.35 . \quad \mathrm{C}_{13} \mathrm{H}_{26} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 44.9 ; \mathrm{H}, 4.8 ; \mathrm{N}, 16.75 \%$ ). The dimethiodide, $\mathrm{m} . \mathrm{p} .168^{\circ}$, was prepared in acetone solution and purified from alcohol-ether (Found : C, 36.45; H, 6.65; N, 5.55. $\mathrm{C}_{15} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{I}_{2}$ requires C, $36.45 ; \mathrm{H}, 6.55 ; \mathrm{N}, 5 \cdot 65 \%$ ).
$1: 4$-Bisdiethylaminopentane.-A solution of $1: 4$-bisdiethylaminopent-2-yne ( 4 g .) in methanol ( 50 c.c.) was shaken with hydrogen and Raney nickel until absorption was complete. After filtration, acidification, and evaporation of the solution, excess of sodium hydroxide ( $20 \%$ ) was added to the residue, and the product isolated with ether. Distillation gave $1: 4$-bisdiethylaminopentane ( 2 g .), b. p. $105^{\circ} / 10 \mathrm{~mm}$., $n_{\mathrm{D}}^{17^{*}} 1 \cdot 4446$ (Found: C, $73 \cdot 2 ; \mathrm{H}, 13 \cdot 95 ; \mathrm{N}, 13 \cdot 2 . \mathrm{C}_{13} \mathrm{H}_{30} \mathrm{~N}_{2}$ requires $\mathrm{C}, 72 \cdot 85 ; \mathrm{H}, 14 \cdot 1$; $\mathrm{N}, 13 \cdot 05 \%$ ). The dipicrate, m. p. 192-194 , crystallised in needles from 2-ethoxyethyl alcohol (Found : $\mathrm{C}, 44 \cdot 65 ; \mathrm{H}, 5 \cdot 4 ; \mathrm{N}, 16 \cdot 65 . \quad \mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 44 \cdot 6 ; \mathrm{H}, 5 \cdot 05 ; \mathrm{N}, 16.9 \%$ ).

1-Piperidino-4-diethylaminopent-2-yne (III; $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}, \mathrm{R}^{\prime \prime} \mathrm{R}^{\prime \prime \prime}=\mathrm{C}_{5} \mathrm{H}_{10}$ ). - 3-Diethylaminobut-l-yne ( 12.5 g .), piperidine ( 8.5 g .), paraformaldehyde ( 4 g .), and dioxan ( $20 \mathrm{c} . \mathrm{c}$.) were heated on the steam-bath for 13 hours, and the product isolated as usual. Distillation gave 1-piperidino-4-diethyl-aminopent-2-yne ( 11.5 g .), b. P. $132-136^{\circ} / 9 \mathrm{~mm} ., n_{\mathrm{D}}^{1{ }^{9}} 1.4810$ (Found: C, $75 \cdot 75 ; \mathrm{H}, 11.5 . \mathrm{C}_{14} \mathrm{H}_{26} \mathrm{~N}_{2}$ requires C, $75 \cdot 6 ; \mathrm{H}, 11 \cdot 8 \%$ ). The dihydrochloride, crystallised from alcohol-ether, had m. p. $200-202^{\circ}$ (decomp.) (Found : C, $56.55 ; \mathrm{H}, 9.35 ; \mathrm{N}, 9.85 . \mathrm{C}_{14} \mathrm{H}_{26} \mathrm{~N}_{2}, 2 \mathrm{HCl}$ requires $\mathrm{C}, 56.95 ; \mathrm{H}, 9.55 ; \mathrm{N}, 9.5 \%$ ). The dipicrate, m. p. 190 - $191^{\circ}$ (decomp.), crystallised in needles from acetone (Found: C, $46 \cdot 1 ; \mathrm{H}, 4 \cdot 6$. $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 45 \cdot 85 ; \mathrm{H}, 4.75 \%$ ). The dimethiodide, m. p. $198-201^{\circ}$ (decomp.), was crystallised from alcohol-ether (Found: C, $37.6 ; \mathrm{H}, 6.0 ; \mathrm{N}, 5.5 . \quad \mathrm{C}_{16} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{I}_{2}$ requires C, 37.95; H , $6.35, \mathrm{~N}, 5.55 \%)$.

1-Piperidino-4-diethylaminopentane.-A solution of 1-piperidino-4-diethylaminopent-2-yne ( $2 \cdot 25 \mathrm{~g}$.) in methanol was shaken with hydrogen and Raney nickel until absorption was complete. Isolation of the product gave l-piperidino-4-diethylaminopentane ( 1.3 g .), b. p. $130-134^{\circ} / 12 \mathrm{~mm}$., $n_{\mathrm{D}}^{17^{\circ}}, 1.4644$ (Found: $\mathrm{N}, 12 \cdot 1 . \quad \mathrm{C}_{14} \mathrm{H}_{30} \mathrm{~N}_{2}$ requires $\mathrm{N}, 12 \cdot 35 \%$ ). The dipicrate, m. p. $177-178^{\circ}$, crystallised as needles from aqueous acetone (Found: C, 45.6; H, 5.25; N, 16.5. $\mathrm{C}_{14} \mathrm{H}_{30} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires C, $45 \cdot 35 ; \mathrm{H}, 5 \cdot 25 ; \mathrm{N}, 16 \cdot 3 \%$ ).

4-Morpholino-1-diethylaminopent-2-yne (III; $\mathrm{RR}^{\prime}=\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}, \quad \mathrm{R}^{\prime \prime}=\mathrm{R}^{\prime \prime \prime}=\mathrm{Et}$ ).-A mixture of 3 -morpholinobut-1-yne ( 13.9 g .), diethylamine ( 8 g .), paraformaldehyde ( 4 g .), and dioxan ( $20 \mathrm{c} . \mathrm{c}$.) was heated on the steam-bath for 15 hours. Isolation of the product gave 4 -morpholino-1-diethylaminopent-$2-y n e\left(18.5 \mathrm{~g}\right.$.), b. p. $86-90^{\circ} / 0 \cdot 1 \mathrm{~mm}$., $n_{\mathrm{D}}^{22^{\circ}} 1 \cdot 4800$ (Found: $\mathrm{N}, 12 \cdot 45 . \mathrm{C}_{13} \mathrm{H}_{24} \mathrm{ON}_{2}$ requires $\mathrm{N}, 12 \cdot 5 \%$ ). The dipicrate, m. p. 204-206 ${ }^{\circ}$, formed needles from glycol monomethyl ether (Found : C, 44.3; H, $4 \cdot 45 ; \mathrm{N}, 16 \cdot 35 . \quad \mathrm{C}_{13} \mathrm{H}_{24} \mathrm{ON}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 44 \cdot 0 ; \mathrm{H}, 4 \cdot 45 ; \mathrm{N}, 16.45 \%$ ).

2: 7-Bisdiethylamino-octa-3:5-diyne (IV; $\mathrm{R}=\mathrm{R}^{\prime}=\mathrm{Et}$ ). - A solution of 3-diethylaminobut-1-yne ( 75 g .) in dilute hydrochloric acid ( $2 \mathrm{~N} ; 310 \mathrm{c} . \mathrm{c}$.) was added to a mixture of cuprous chloride ( 15 g .), ammonium chloride ( 45 g. ), and water ( $53 \mathrm{c} . \mathrm{c}$. ). The resulting solution was warmed to $50-55^{\circ}$ and stirred vigorously, and air was bubbled through it until green salts began to be deposited ( 9 hours). The mixture was cooled and filtered, and excess of concentrated ammonia solution added to the filtrate. The precipitated oil was extracted with ether, the ethereal solution evaporated, and the residue distilled, giving 2: 7-bisdiethylamino-octa-3:5-diyne ( 68 g .), b. p. $130-134^{\circ} / 0 \cdot 2 \mathrm{~mm}$., $n_{D}^{18^{\circ}} 1.4958$ (Found: C, $77 \cdot 35 ; \mathrm{H}, 11 \cdot 2 ; \mathrm{N}, 11 \cdot 25 . \quad \mathrm{C}_{16} \mathrm{H}_{28} \mathrm{~N}_{2}$ requires $\mathrm{C}, 77 \cdot 3 ; \mathrm{H}, 11 \cdot 35 ; \mathrm{N}, 11 \cdot 3 \%$ ). The dipicrate, $\mathrm{m} . \mathrm{p} .230^{\circ}$
(decomp.) (bath preheated to $190^{\circ}$ ), crystallised as leaflets from 2-ethoxyethyl alcohol (Found : C, 47.8; $\mathrm{H}, 4.85$; $\mathrm{N}, 16.3 . \quad \mathrm{C}_{16} \mathrm{H}_{28} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 47.6 ; \mathrm{H}, 4.85 ; \mathrm{N}, 15.9 \%$ ).

2:7-Bisdiethylamino-octane.-2:7-Bisdiethylamino-octa-5:7-diyne ( $4 \cdot 5 \mathrm{~g}$. ) in methanol ( 50 c.c.) was hydrogenated (Raney nickel), giving 2: 7-bisdiethylamino-octane ( 2 g.), b. p. $83-87^{\circ} / 0 \cdot 1 \mathrm{~mm}$., $n_{\mathrm{D}}^{20^{\circ}}$ 1.4612 (Found: N, $11 \cdot 15$; equiv., $128 \cdot 2 . \quad \mathrm{C}_{16} \mathrm{H}_{36} \mathrm{~N}_{2}$ requires $\mathrm{N}, 10 \cdot 9 \%$; equiv., $128 \cdot 2$ ). The dipicrate, m. p. 186- $190^{\circ}$ (decomp.), crystallised from 2-ethoxyethyl alcohol as leaflets (Found : C, $47 \cdot 2$; H, $5 \cdot 6$. $\mathrm{C}_{16} \mathrm{H}_{36} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 47 \cdot 05 ; \mathrm{H}, 5 \cdot 9 \%$ ).

2:7-Dimorpholino-octa-3:5-diyne (IV; $\quad \mathrm{RR}^{\prime} \doteq \mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}$ ).—3-Morpholinobut-1-yne (25 g.) was dissolved in hydrochloric acid ( $2 \mathrm{~N} ; 85 \mathrm{c} . \mathrm{c}$.), and the solution added to a mixture of cuprous chloride ( 4 g .), ammonium chloride ( 12 g .), and water ( $14 \mathrm{c} . \mathrm{c}$.). Aerial oxidation (for 5.5 hours) and isolation as in the case of the diethylamino-analogue gave 2:7-dimorpholino-octa-3:5-dyne (22 g.), which crystallised from alcohol in prisms, m. p. $114-115^{\circ}$ (Found: C, $69 \cdot 4 ; \mathrm{H}, 8 \cdot 4 ; \mathrm{N}, 9 \cdot 9 . \mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{~N}_{2}$ requires C, 69.55 ; H, 8.75 ; N, $10.15 \%$ ). The dipicrate, m. p. $224-227^{\circ}$ (decomp.), crystallised in prisms from 2-ethoxyethyl alcohol (Found : C, $45 \cdot 8 ; \mathrm{H}, 4 \cdot 3 . \quad \mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires C, $45 \cdot 75 ; \mathrm{H}, 4 \cdot 1 \%$ ). The dihydrochloride, leaflets from aqueous acetone, had m. p. $242^{\circ}$ (decomp.) (Found : N, 8.15. $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{~N}_{2}, 2 \mathrm{HCl}$ requires $\mathrm{N}, 8 \cdot 0 \%$ ).

2:7-Di isopropylamino-octa-3:5-diyne (IV; $\mathrm{R}=\mathrm{Pr}^{\mathrm{i}} ; \quad \mathrm{R}^{\prime}=\mathrm{H}$ ).—A solution of 3-isopropyl-aminobut-1-yne ( 100 g .) in dilute hydrochloric acid ( $2 \mathrm{~N} ; 440 \mathrm{c} . \mathrm{c}$.) was added to a mixture of cuprous chloride ( 20 g .), ammonium chloride ( 60 g .), and water ( $70 \mathrm{c.c}$.), and the oxidation carried out as usual. Isolation gave 2:7-diisopropylamino-octa-3: 5-diyne monohydrate ( 105 g. ), m. p. $61^{\circ}$, which crystallised as needles from aqueous methanol (Found: C, 70.7 ; $\mathrm{H}, 10.95 ; \mathrm{N}, 11.95 . \mathrm{C}_{14} \mathrm{H}_{24} \mathrm{~N}_{2}, \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 70.5 ; \mathrm{H}, 11.0 ; \mathrm{N}, 11.75 \%$ ). On standing in a desiccator this solid gives an oil, presumably the anhydrous diamine, which on exposure to the atmosphere is rapidly reconverted into the original hydrate. The dipicrate, m. p. 198-200 (decomp.), crystallised in prismatic needles from aqueous methanol (Found: C, $45.85 ; \mathrm{H}, 4.4 ; \mathrm{N}, 16.7 . \mathrm{C}_{14} \mathrm{H}_{24} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires C, $46 \cdot 05 ; \mathrm{H}, 4.45 ; \mathrm{N}, 16.5 \%$ ). The dihydrochloride, purified from alcohol-ether, had m. p. $242.5-243.5^{\circ}$ (Found : C, $56.95 ; \mathrm{H}, 8.7$; $\mathrm{N}, 9 \cdot 25 . \mathrm{C}_{14} \mathrm{H}_{24} \mathrm{~N}_{2}, 2 \mathrm{HCl}$ requires C, $57.35 ; \mathrm{H}, 8 \cdot 95 ; \mathrm{N}, 9.55 \%$ ).

2:7-Diisopropylamino-octane.-A solution of 2:7-diisopropylamino-octa-3:5-diyne monohydrate ( 92 g.) in methanol ( 800 c.c.) was hydrogenated over Raney nickel until absorption was complete. Isolation of the product gave 2:7-diisopropylamino-octane ( 71 g .), b. p. $92-95^{\circ} / 0 \cdot 5 \mathrm{~mm} ., n_{\mathrm{D}}^{20^{\circ}} 1 \cdot 435$ (Found : $\mathrm{N}, 12 \cdot 2, \quad \mathrm{C}_{14} \mathrm{H}_{32} \mathrm{~N}_{2}$ requires $\mathrm{N}, 12 \cdot 25 \%$ ). The dihydrochlovide, crystallised from alcohol-ether, had m. p. $270^{\circ}$ (decomp.) (Found : C. 55.65 ; $\mathrm{H}, 11 \cdot 25 . \quad \mathrm{C}_{14} \mathrm{H}_{32} \mathrm{~N}_{2}, 2 \mathrm{HCl}$ requires $\mathrm{C}, 55 \cdot 8 ; \mathrm{H}, 11 \cdot 35 \%$ ).

2:7-Di-n-butylamino-octa-3:5-diyne (IV; $\mathrm{R}=\mathrm{Bu} ; \mathrm{R}^{\prime}=\mathrm{H}$ ).-A solution of $3-n$-butylaminobut1 -yne ( 11 g .) in dilute hydrochloric acid ( $2 \mathrm{~N} ; 45 \mathrm{c.c}$.) was added to a mixture of cuprous chloride ( 2 g .), ammonium chloride ( 6 g .), and water ( $7 \mathrm{c} . \mathrm{c}$.). Aerial oxidation ( 14 hours) at $50^{\circ}$ and isolation of the product gave 2: 7-di-n-butylamino-octa-3:5-diyne ( $7 \cdot 2 \mathrm{~g}$.), b. p. $128-132^{\circ} / 0 \cdot 06 \mathrm{~mm}$., $n_{\mathrm{D}}^{2{ }^{\bullet}} 1 \cdot 4952$ (Found : $\mathrm{N}, 11 \cdot 4$. $\quad \mathrm{C}_{16} \mathrm{H}_{28} \mathrm{~N}_{2}$ requires $\mathrm{N}, 11.3 \%$ ). The di-a-naphthylurea, m. p. $163-164^{\circ}$, crystallised in micro-needles from aqueous alcohol (Found : C, $77 \cdot 7 ; \mathrm{H}, 7 \cdot 1 ; \mathrm{N}, 9 \cdot 25$. $\mathrm{C}_{38} \mathrm{H}_{42} \mathrm{O}_{2} \mathrm{~N}_{4}$ requires $\mathrm{C}, 77 \cdot 75$; H, $7 \cdot 2$; $\mathrm{N}, 9.55 \%$ ).
$1: 10$-Bisdiethylaminodeca-2:8-diene-4:6-diyne (V).—A solution of 1-diethylaminopent-2-en-4-yne ( 4 g. ; prepared from pent-2-en-4-yn-l-ol by chlorination with thionyl chloride and then treatment of the chloride with diethylamine, cf. Jones, Lacey, and Smith, $J ., 1946,940$ ) in dilute hydrochloric acid ( $2 \mathrm{~N}, 14 \mathrm{c} . \mathrm{c}$.) was added to a mixture of cuprous chloride ( 0.65 g .), ammonium chloride ( 1.9 g .) , and water ( $2.25 \mathrm{c} . \mathrm{c}$.). Aerial oxidation ( $5 \cdot 5$ hours) and isolation of the product gave bisdiethylaminodeca-2 : 8-diene-4:6-diyne ( $2 \cdot 6$ g.), b. p. $155-160^{\circ} / 0 \cdot 1 \mathrm{~mm}$., $n_{\mathrm{D}}^{25^{\circ}} 1 \cdot 5630$ (Found: C, $78 \cdot 6 ; \mathrm{H}, 9 \cdot 8 . \quad \mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{2}$ requires $\mathrm{C}, 79.3 ; \mathrm{H}, 10.3 \%$ ). The dipicrate, $\mathrm{m} . \mathrm{p} .146-148^{\circ}$, crystallised in prisms from aqueous acetone (Found : C, $49 \cdot 0 ; \mathrm{H}, 4 \cdot 85$; N, $15 \cdot 3 . \quad \mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{2}, 2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~N}_{3}$ requires $\mathrm{C}, 49 \cdot 3 ; \mathrm{H}, 4 \cdot 7$; N , $15.35 \%$ ). The dimethiodide, needles from alcohol, had m. p. $211^{\circ}$ (decomp.) (Found : C, 43.2; H, $6 \cdot 1$; $\mathrm{N}, 4.85 . \quad \mathrm{C}_{20} \mathrm{H}_{34} \mathrm{~N}_{2} \mathrm{I}_{2}$ requires $\mathrm{C}, 43.15 ; \mathrm{H}, 6.15 ; \mathrm{N}, 5.05 \%$ ). The dihydrochloride, $\mathrm{m} . \mathrm{p} .215^{\circ}$ (decomp.), crystallised as needles from alcohol (Found: $\mathrm{N}, 8 \cdot 25 . \mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{2}, 2 \mathrm{HCl}$ requires $\mathrm{N}, 8 \cdot 1 \%$ ). Light absorption in methanol : maxima 2380, 2490, 2775, 2970, 3160; $\varepsilon_{\text {max }}$. $28,500,26,000,9,500,16,500$, 27,000 and 23,000 respectively.

Diacetyl Dianil (VIII).-A mixture of aniline ( 45 g .), mercuric oxide ( 15 g .), and boron trifluorideether complex ( 6 g .) was warmed to $50^{\circ}$, and 3 -diethylaminobut-1-yne ( 60 g .) added dropwise ( 0.5 hour) with rapid stirring. The mixture was stirred at $50^{\circ}$ for a further 4 hours, cooled, and filtered. Distillation of the filtrate gave 3-diethylaminobut-1-yne ( 30 g. ), aniline ( 15 g .), and a residue which on trituration with methanol solidified. Recrystallisation from the same solvent gave diacetyl dianil ( 17 g .) as yellow plates, m. p. $138^{\circ}$ undepressed on admixture with an authentic specimen (von Pechmann, Ber., 1888, 21, 1415, gives m. p. $139^{\circ}$ ) (Found: $\mathrm{N}, 12 \cdot 0$. Calc. for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2}$ : N, $11 \cdot 85 \%$ ).

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